Aldehyde oxidase (E.C.1.2.3.1) (AO) belongs to the small family of molybdoflavoproteins that includes xanthine oxidoreductase (XOR) and sulphate oxidase1. AO is a cytosolic enzyme that contains FAD, molybdenum and iron-sulfur centers, and is closely related to xanthine oxidase. The enzyme can catalyze the oxidation of number of aldehydes and nitrogenous heterocyclic compounds. Moreover, the enzyme in presence of an electron donor can also mediate the concomitant reduction of a variety of compound such as sulfoxides, N-oxides, nitrosamine, hydroxamic acids, azo dyes, epoxides, aromatic nitro compounds and 1,2-benzisoxazole derivatives1. AO possesses wider substrate specificity as compared to xanthine oxidase and plays an important role in the metabolism of xenobiotics as well as endobiotics2. AO is an important Phase I oxidative enzyme that contributes to the metabolism of heterocyclic structures that possess an electropositive carbon centre (generally C adjacent to a heteroatom like N, S, or O)3. Recently, AO has been suggested to play an important role in visual processes, since the enzyme catalyzes the biotransformation of the principal component of the visual pigment all trans-retinal to retinoic acid, the active form of vitamin A3. Mammalian liver AO is also responsible for metabolism of nicotine metabolite (S)-nicotine Δ1,5-iminium ion, to (S)-cotinine in rat; the α-N-oxidation of brominidine to its 2, 3-dioxoderivatives, major metabolites in humans4. The metabolism of tamoxifen 4-aldehyde to the corresponding carboxylic acid in rat, activation of pyrimidine prodrug nucleus to uralic acid in mouse, rabbit and rat and the conversion of famciclovir to penciclovir in humans are all catalyzed by AO4.

Our laboratory has been involved in the establishment of both enzyme fraction isolation protocols and enzyme assay protocols for various important drug metabolizing enzymes. The present study was initiated for isolation of crude fraction of AO from various animal species and determination of kinetic parameters of AO for substrate benzaldehyde by spectrophotometric method.

Animal livers were obtained from the Department of Pharmacology, Bombay College of Pharmacy, Mumbai. The animals and the strains used were as follows; rat (Charles Foster), mouse (Swiss Albino), guinea pig (Albino) and rabbit (New Zealand). The animals used in this study were those that were sacrificed as part of other experiments approved by the Institutional Animal Ethics Committee. It should be noted that the potential of these experiments to alter liver function (and consequently AO content) although a possibility, was not taken into consideration since the intention of this study was to only evaluate the AO isolation.
procedure and AO activity determination. The livers obtained were either used fresh or stored at -70°C until use. Potassium dihydrogen orthophosphate, dipotassium hydrogen orthophosphate, benzaldehyde and EDTA were purchased from S. D. Fine Chem. Ltd., Mumbai. Bradford’s macro method protein estimation kit was purchased from Bangalore Genie Pvt. Ltd., Bangalore. Ammonium sulphate was purchased from Himedia Lab., Mumbai. All other chemicals and reagents used in the study were of AR grade.

Crude fractions of AO were isolated from animal livers by previously reported procedures. Briefly, 10 g of liver was homogenized in 5 volumes of 0.05 M potassium phosphate buffer, pH 6.8, for 5 min in a Potter glass homogenizer equipped with a Teflon pestle. The homogenate was then rapidly heated to 55°C on a water bath, maintained at this temperature for 5 min, and then cooled quickly to below 10°C in an ice bath. During both the heating and cooling steps the homogenate was stirred. The heat-treated and cooled homogenate was centrifuged at 16 000×g for 15 min and the precipitate discarded. Solid ammonium sulphate was added to the supernatant to a final concentration of 60% saturation (37.56 g/100 ml), the mixture centrifuged at 20 000×g for 15 min and the pellet was suspended in 10 ml of 0.05 M potassium phosphate buffer, pH 7.8, containing 0.005% EDTA and stored at -70°C for further use. Three different samples of AO fractions were obtained from different liver samples of each animal and all of the subsequent estimations were done in duplicate for each of the three isolated AO fractions per animal.

AO activity assay was performed according to the reported method using ‘blank reversal technique’. A Shimadzu spectrophotometer (UV 160 A) with matched 10 mm quartz cuvettes was used for the assay. The assay mixture contained 50 mM potassium phosphate buffer, pH 7.8, containing 0.005% EDTA, 300 μl of 0.01 M potassium ferricyanide solution, 50 μl of 0.3 M benzaldehyde and 50 μl of enzyme sample in final volume of 3 ml, in a 10 mm quartz cuvette. The blank cuvette consisted of all the components listed above except enzyme. AO activity determination was initiated by addition of enzyme and was monitored by analyzing the amount of potassium ferricyanide reduced at 420 nm at ambient temperature. The velocity of reaction was determined as μmol of potassium ferricyanide reduced per ml per min using extinction coefficient (ε) of 2080 M⁻¹ cm⁻¹. One unit of enzyme activity was defined as the amount of enzyme that catalyzed the reduction of 1 μmol of potassium ferricyanide per min under the stated conditions.

Protein concentrations of partially purified AO fractions were determined spectrophotometrically according to the Bradford method with Bradford macro method kit, using bovine serum albumin as standard, according to the manufacturer’s instructions.

K_m (Michaelis-Menten constant) and V_max (maximum velocity) values for the oxidation of benzaldehyde to benzoic acid with rabbit and guinea pig liver AO fractions were determined. Initial experiments to determine linearity of reaction with respect to both enzyme amount and time were conducted (data not shown). Product formation was kept to 10% or below for adherence to Michaelis-Menten assumptions. The assay methodology used was similar to described above for enzyme assay, except that different concentrations of benzaldehyde i.e. 5, 10, 15, 20, 25, 50, 100, and 500 μM were used. The velocity of reaction was determined as μmol of potassium ferricyanide reduced per ml per min using extinction coefficient (ε) of 2080 M⁻¹ cm⁻¹. K_m and V_max values were determined using the Lineweaver-Burk, Eadie-Hofstee and Hanes plotting methods. The line of best fit through the points on the plot was determined using linear regression by least squares method using Microsoft Excel (Microsoft Office XP).

Unit activity and specific activity of AO fractions from different animal liver cytosolic fractions were measured spectrophotometrically, using benzaldehyde as the specific probe substrate for AO. The mean unit AO activities and specific AO activities are listed in Table 1. AO activity was present in rabbit and guinea pig isolated liver AO fractions. AO activity was however absent in rat and mouse liver AO fractions. Further, enzyme kinetic parameter estimations were therefore done only with rabbit and guinea pig liver AO fractions. K_m and V_max values for the oxidation of benzaldehyde to benzoic acid by aldehyde oxidase fractions were measured for rabbit and guinea pig liver AO liver fractions. Mean values of K_m and V_max of AO as determined using three different plotting methods viz. Lineweaver-Burk plot, Eadie-Hofstee plot and Hanes plot are listed in Table 2.

In the present study, high aldehyde oxidase activity was detected in rabbit and guinea pig liver whereas,
AO activity was absent in rat and mouse livers. This was expected as previous studies have indicated that little or no AO is present in rat and mouse livers. Such species differences are known for AO activity, for example, high AO activity is observed in rabbit liver while AO activity is undetectable in dogs.

The kinetic parameters i.e. $K_m$ and $V_{max}$ were determined using three Lineweaver-Burk plot, Eadie-Hofstee plot and Hanes plot. Of the three plotting methods, it is suggested that the Eadie-Hofstee plot is superior to the other two methods of plotting data, due to the inherent stringency of this plotting method. The reported $K_m$ value for benzaldehyde oxidation by AO from guinea pig and hamster liver AO is 19 $\mu$M and 11 $\mu$M for human liver AO. The results obtained in this study shows that the $K_m$ values as determined by Lineweaver-Burk plot, for rabbit and guinea pig liver AO fractions were 10.7 ± 3.62 $\mu$M and 19.1 ± 1.67 $\mu$M, respectively, which are in reasonable agreement with the reported values. The values of $K_m$ as determined by different plotting methods are also in the same range (10.3-19.1 $\mu$M).

Overall, this study presents the evaluation of a simple method for the isolation of crude fractions containing AO from different animal livers and a spectrophotometric method for the determination of AO enzyme kinetic parameters in crude fractions, with benzaldehyde as the model substrate. This assay is of utility is the establishment of drug metabolism study protocols in drug metabolism research.
ACKNOWLEDGEMENTS

The authors would like to extend special thanks to Dr. M. L. Kubal, Wockhardt Ltd., Aurangabad for providing HPLC columns used in this work. The authors also wish to thank AICTE (File No. 8019/RDII/R and D/PHA (213) 2000-01) and DST-“FIST Program”-2003 [485] for providing the research grants for this work.

REFERENCES


Microwave-Induced Synthesis of Schiff Bases of Aminothiazolyl Bromocoumarins as Antibacterials

K. N. VENUGOPALA* AND B. S. JAYASHREE1
Department of Pharmaceutical Chemistry, Al-Ameen College of Pharmacy, Near Lalbagh Main Gate, Hosur Road, Bangalore - 560 027, 1Department of Pharmaceutical Chemistry, Manipal College of Pharmaceutical Sciences, MAHE, Manipal - 576 104, India

Venugopala, et al.: Microwave Synthesis of Aminothiazolyl Bromocoumarins

A fast and highly efficient method for the synthesis of some of the schiff bases of aminothiazolylbromocoumarin (4a-m) has been performed by microwave irradiation of 2’-amino-4’-(6-bromo-3-coumarinyl) thiazole (3) and substituted aromatic aldehydes (a-m). Microwave assisted reactions have resulted in better yields of the desired products than when prepared under conventional conditions. The resulting products were evaluated for their qualitative and quantitative antibacterial activity.

Key words: Bromocoumarin, microwave, characterization, antibacterials

The synthesis of coumarins and their medicinal chemists has attracted the attention of organic and medicinal chemists as these are widely used as fragrances, pharmaceuticals and agrochemicals1. Benz-2pyrines and its heterocyclic derivatives, in particular schiff bases and carboxamides of 3-thiazoyl substituted coumarins, display important biological properties such as analgesic, anti-inflammatory2,3, anticoagulant4, antimicrobial, antiviral5 and HIV