Nephroprotective Effects of *Berberis Vulgaris* L. Total Extract on Lead Acetate-induced Toxicity in Mice

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Laamech, et al. *Berberis vulgaris* Effect on Lead Nephrotoxicity in Mice

This study aimed to investigate the potential effects of *Berberis vulgaris* L. against lead acetate-induced nephrotoxicity in Swiss albino mice. Mice were exposed chronically to lead acetate, at a dose of 5 mg/kg, alone or in conjunction with *B. vulgaris* extract (25, 50, 100 and 150 mg/kg). Lead acetate administration increased lead burden in blood and kidneys, altered renal biomarkers (serum creatinine, urea and uric acid), increased malondialdehyde and protein carbonyls levels, and decreased reduced glutathione, glutathione peroxidase, catalase, and superoxide dismutase. Histological studies showed glomerular degeneration and hypercellularity. However, *B. vulgaris* extract co-treatment significantly and dose-dependently restored the renal parameters, antioxidant enzymes, and histopathological changes near to the normal status, mainly via its antioxidant properties and/or by preventing lead bioaccumulation. Therefore, *B. vulgaris* extract showed nephroprotective effects against lead-induced nephrotoxicity mainly through its antioxidant and metal chelating properties.

Key words: Lead acetate, *Berberis vulgaris*, mice, nephrotoxicity, antioxidant enzymes, lipids peroxidation

Many heavy metals exist in our environment both naturally and from pollution. Some of them are very toxic and ranked as human carcinogens. Accordingly, lead (Pb) is a systemic toxic metal known for multiple industrial, domestic, agricultural, medical and technological applications that contribute to its wide distribution in the environment. Exposure to lead occurs mainly through intake of lead-contaminated food, water, dusts and paints[1-2]. This triggers generation of reactive oxygen species (ROS), and depletes the cellular antioxidant capacity. An imbalance of antioxidant pool affects cellular organelles, antioxidant enzymes, and damages membranes, DNA, proteins, and finally destroys the tissues[3-4]. Therefore, exogenous administration of antioxidant substances would have a beneficial effect on the cells’ antioxidant system to combat lead intoxication. In accordance, there are growing interests in using natural compounds to treat lead nephrotoxicity[5-6].

*Berberis vulgaris* L. (Berberidaceae) is a medicinal plant widely distributed in the Atlas and Rif mountains of Morocco. The species is commonly known as āğrys in Moroccan pharmacopoeia, and used to cure renal disorders and other diseases[7]. Moreover, it is the most widely used drug in homeopathic system of medicine for kidney pain as well as removal of kidney stones[8]. Many studies report protective effects of *B. vulgaris* and some of its bioactive molecules, especially berberine, which is the most important alkaloid claimed to be responsible for beneficial effects of *B. vulgaris*. Moreover, many studies report antioxidant effects of *B. vulgaris*, and several polyphenols with antioxidant activities have been isolated from root, bark, leaf and fruit[8-9]. To our knowledge, this is the first study to evaluate *B. vulgaris* effects against lead-induced nephrotoxicity in mice. Therefore, this study aims to investigate the potential protective effects of *B. vulgaris* against kidney damage induced by subchronic administration of lead acetate.

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MATERIALS AND METHODS

Lead acetate (PbAc), formalin (10%), ethanol (90%), nitric acid (HNO₃, 65%), Triton X-100 (0.2%), sulfoisalicylic acid (4%), Ellman’s reagent (dithionitrobenzene, DTNB), 0.1 M Tris–HCl, 0.001 M EDTA buffer (pH 7.4), 5% TCA, thiobarbituric acid reagent (TBA, 0.67%), 1,1,3,3-tetraethoxypropane-2,4-dinitrophenyl hydrazine (DNPH), ascorbic acid (1%), palladium chloride (PdCl₂), toluene, paraffin, hematoxylin and eosin solutions. All reagents and chemical products used in this study were of analytical grade and were purchased from Sigma Chemical Co. (St. Louis, France).

Plant material and preparation of the lyophilized aqueous extract:

Samples of *B. vulgaris* were collected in April-May, 2013 from mountain areas of Imouzzer Marmoucha (1713 m, 33° 28' 37 N, 4° 17' 44 W), a province of Fez-Boulemane region. Mature whole plant was authenticated at the herbarium of the Scientific National Institute (Rabat), where a voucher specimen (C3) is kept. The BV stem bark was separated and frozen until extraction. The plant collection was conducted in accordance with the ethical standards outlined in the international, national and institutional rules concerning the biodiversity rights.

Fifty grams of dried powder of *B. vulgaris* stem bark was decocted with 500 ml of water. The mixture was heated and boiled under reflux for 30 min. The decoction obtained was centrifuged, filtered, frozen at −20°C, and then lyophilized (FreeZone® Dry 4.5, USA). This yielded a total of 7.5 g of extract (15% w/w).

Sixty adult Swiss albino mice (25-30 g) were used in this study. The mice were kept in standard polypropylene cages. Animals were maintained under standard laboratory conditions of temperature (25±2°C), relative humidity (50±15%), 12 h light-dark cycle, standard diet and water *ad libitum*. The care and handling of the animals were in accordance with the internationally accepted standard guidelines for use of animals, and the protocol was approved by our institutional committee on animal care following the French Technical Specifications for the Production, Care and Use of the Laboratory Animals.

Experimental procedure:

After 2 w of acclimatization to the laboratory conditions, the animals were randomly divided into six groups of 10 mice each and treated orally by force-feeding as follows[10]: Group 1: received double distilled water as vehicle during the whole course of study; served as normal control (CT); Group 2, received lead acetate (5 mg/kg/day) dissolved in double distilled water for 40 days; served as toxic control (PbAc); Groups 3-6 received *B. vulgaris* aqueous extract at doses of 25, 50, 100 and 150 mg/kg, respectively, once daily for 30 days from 11 days after beginning of lead acetate exposure (5 mg/kg/day) to the end of the experiment.

The *B. vulgaris*-aqueous extract doses were chosen according to previous studies on the plant or on its main constituent (berberine), and on the basis of acute toxicity study of the plant[11-12]. After 42 days, the animals were given rest overnight, and then on the next day they were sacrificed by cervical dislocation under light ether anesthesia. Before sacrificing the animals, blood samples were collected from retro-orbital venous plexus in Eppendorf tubes rinsed with EDTA anticoagulant for lead bioaccumulation assay (5 mice from every experimental group), and other tubes without anticoagulant were used to determine biochemical parameters. The blood was centrifuged at 2200 g for 15 min at 4°C, serum samples were drawn and kept at −20°C until biochemical assays (5 mice from every experimental group). The kidneys were excised, cleaned and washed with ice cold normal saline. The portions of kidneys were homogenized in 0.1 M Tris HCl–0.001 M EDTA buffer (pH 7.4) and centrifuged at 12 000 g for 30 min at 4°C. The supernatant was collected and used for determining biochemical parameters (n=5). Other kidneys were kept in 10% formalin (n=5) until histological assay.

Estimation of blood and kidneys lead burden:

The accumulation of lead in blood (Pb-B) and kidneys (Pb-K) has been measured by flame atomic absorption spectroscopy[13].

Serum creatinine, urea and uric acid levels:

Serum creatinine, urea and uric acid levels were estimated by spectrophotometry using Commercial kits Biomaghreb Diagnostics Casablanca, Morocco (refs. 20043, 20147, 20154; respectively).

Reduced glutathione (GSH):

Kidney GSH content was determined by Ellman’s method[14], modified by Jollow *et al.*[15], based on the development of a yellow color when DTNB is added to compounds containing sulphydryl groups. Briefly, 3 ml of sulfoisalicylic acid (4%) were added to 500
ml of homogenate tissues for deproteinization. The mixture was centrifuged at 2500 g for 15 min. Then Ellman’s reagent was added to 500 ml of supernatant. The absorbance was measured at 412 nm after 10 min. Total GSH content was expressed as µmol/g of tissue.

**Glutathione peroxidase (GPx) activity:**

GPx was measured according to Flohe and Gunzler[16]. The enzyme activity was expressed as nanomoles of GSH oxidized/min/mg protein.

**Catalase activity:**

Catalase (CAT) activity was assayed by the decomposition of hydrogen peroxide according to the method of Aebi[17]. Decrease in absorbance due to H₂O₂ degradations was monitored at 240 nm for 1 min and the enzyme activity was expressed as µmol H₂O₂ consumed/min/mg protein.

**Superoxide dismutase (SOD) activity:**

The total SOD activity was evaluated by measuring the inhibition of pyrogallol activity[18]. One unit (U) corresponds to the enzyme activity required to inhibit half of the oxidation of pyrogallol. Then SOD activity was expressed as U/mg of protein.

**Assay of lipid peroxidation:**

Lipid peroxidation in the renal tissue was estimated calorimetrically by measuring thiobarbituric acid reactive substances (TBARS) which were expressed in terms of malondialdehyde content[19]. Briefly, Aliquots of kidney homogenates were mixed with 1ml of 5% TCA and centrifuged at 4000 g for 10 min. one milliliter of thiobarbituric acid reagent (TBA, 0.67%) was added to 500 ml of supernatant and heated at 95° for 15 min. The mixture was then cooled and was measured for absorbance at 532 nm. The MDA values were calculated using 1,1,3,3-tetraethoxypropane as the standard and expressed as nanomoles of MDA/g of tissue.

**Protein carbonyl assays:**

Protein oxidation was determined based on the reaction of the carbonyl groups with 2,4-dinitrophenylhydrazine (DNPH) to form 2,4-dinitrophenylhydrazone, as described by Reznick and Packer[20]. Samples were read at 370 nm and carbonyl content was calculated using the molar absorption coefficient for aliphatic hydrazones (22 000 M⁻¹cm⁻¹) and expressed as nmol carbonyl/mg protein.

**Histopathological study of kidney:**

For qualitative analysis of kidney histology, the tissue samples were fixed for 48 h in 10% formalin solution, dehydrated in ascending graded series of ethanol, cleared in toluene and embedded in paraffin. Sections of the tissue (5-6 mm thickness) were prepared by using a rotary microtome and stained with hematoxylin and eosin (H and E) for microscopic observations.

**Statistical analysis:**

The data were analyzed using the statistical package program GraphPad Prism 5.03, USA. Data were statistically calculated by utilizing one-way ANOVA and expressed as mean±standard error of the mean (SEM) followed by Tukey’s test. Moreover, Student’s unpaired t-test was used when comparison between two groups was required. The values were considered significant when P<0.05.

**RESULTS AND DISCUSSION**

Before treatment, the initial body weight of mice was similar in all animals groups. A non-significant decrease in the level of the final body weights (BW) was observed in PbAc-treated rats when compared with control rats (CT). Administration of *B. vulgaris* at doses of 100-150 mg/kg body weight along with lead significantly (P<0.05) raised the values of final body weight when compared to normal control group. Likewise, *B. vulgaris* treatments increased significantly and dose-dependently (P<0.05) body weight in comparison with toxic control group.

Kidneys weight values were not significantly lowered in toxic control group as compared to normal control group. However, *B. vulgaris*-treated groups depicted significant (P<0.05) and dose-independent weight intake in comparison with toxic control group.

Mice exposure to PbAc significantly (P<0.05) raised the bioaccumulation of lead in blood and kidneys when compared with control group. Compared to Pb-acetate treated group, *B. vulgaris* extract decreased significantly (P<0.05) at any dose tested the blood and kidney lead burden (Table 1).

Daily treatment with lead resulted in significant (P<0.05) increase in serum levels of creatinine and urea in PbAc-treated group when compared to normal control (+64% and +75%, respectively). Co-administration of *B. vulgaris*-aqueous extract at doses of 100-150 mg/kg decreased significantly (P<0.05) the levels of these renal markers when compared to the
### Table 1: Body and Kidneys Weights, Blood and Kidney Lead Burden of Mice

<table>
<thead>
<tr>
<th>Parameters</th>
<th>CT</th>
<th>PbAc</th>
<th>PbAc+25</th>
<th>PbAc+50</th>
<th>PbAc+100</th>
<th>PbAc+150</th>
</tr>
</thead>
<tbody>
<tr>
<td>Initial body weight (g)</td>
<td>28.33±0.67</td>
<td>28.26±0.52</td>
<td>27.79±0.29</td>
<td>27.73±0.66</td>
<td>28.37±0.4</td>
<td>27.67±0.42</td>
</tr>
<tr>
<td>Final body weight (g)</td>
<td>30.5±0.81</td>
<td>28.55±0.56</td>
<td>31.18±0.36</td>
<td>31.39±0.27</td>
<td>32.77±0.44</td>
<td>32.8±0.25</td>
</tr>
<tr>
<td>Kidneys weight (g)</td>
<td>0.45±0.003</td>
<td>0.43±0.01</td>
<td>0.46±0.003</td>
<td>0.46±0.004</td>
<td>0.46±0.004</td>
<td>0.46±0.01</td>
</tr>
<tr>
<td>Pb-B (µg/ml)</td>
<td>0.009±0.00</td>
<td>0.18±0.01</td>
<td>0.16±0.01</td>
<td>0.09±0.01</td>
<td>0.04±0.02</td>
<td>0.02±0.01</td>
</tr>
<tr>
<td>Pb-K (µg/g wet tissue)</td>
<td>0.009±0.00</td>
<td>0.17±0.01</td>
<td>0.08±0.003</td>
<td>0.03±0.004</td>
<td>0.03±0.003</td>
<td>0.02±0.003</td>
</tr>
</tbody>
</table>

Body and Kidneys weights, blood and kidney lead burden of mice control group (CT), lead acetate-treated mice (PbAc), and *Berberis vulgaris* treated mice (0, 25, 50, 100, 150 mg/kg) in combination with PbAc (PbAc+BV). Values are expressed as mean±SEM for five animals in each group. *Values differ significantly (P<0.05) from normal control group (CT)." Values differ significantly (P<0.05) from toxic control group (PbAc)."

Daily dosing with lead induced a significant decrease of kidney GSH content (-60%; P<0.05) in PbAc group compared to normal control. A significant and dose-dependent recovery (P<0.05) in kidney GSH content was noticed in both 100 and 150 mg/kg-treated groups in comparison with toxic control group (Table 3). Glutathione peroxidase (GPx), catalase (CAT) and superoxide dismutase (SOD) activities were significantly decreased (about -60%) in kidney tissue of lead-treated mice as compared to their respective control group (P<0.05). Co-treatment of lead-treated mice with *B. vulgaris* extract, at doses higher or equal to 100 mg/kg, improved significantly the activities of GPx and CAT in kidneys of *B. vulgaris*-treated mice to a level similar to that of the normal control group. Whereas, the renal SOD activity was restored significantly to normal status by co-administration of BV at a dose of 150 mg/kg as compared to the toxic control group (PbAc) (Table 3).

PbAc exposure caused a significant increase (P<0.05) in the level of lipid peroxidation (+64%) and protein carbonyls (+96%) in the kidney tissue. Co-administration of *B. vulgaris* extract at doses of 100 and 150 mg/kg significantly (P<0.05) lowered the levels of MDA and PCO in comparison with their respective control values, respectively (Table 3).

The histological examination of the kidney tissues of control mice showed normal cyto-architecture of glomerulus and tubules structure (fig. 1a). Mice treated with PbAc alone showed signs of glomerular degeneration as glomerular hypercellularity (fig. 1b), while those treated with lead-acetate in combination with *B. vulgaris* extract (150 mg/kg) restored more or less normal glomerular structures of renal tissues (fig. 1c-f).

Chronic exposure to lead is known to induce a broad spectrum of toxicological effects and biochemical dysfunctions including those of the kidney[21-22]. Although the precise mechanism of lead toxicity is not entirely clear, it is evidenced that lead exposure generates reactive oxygen metabolites and causes oxidative stress in living systems[3-4]. Moreover, natural antioxidants can combat the over-production of free radicals and activated oxygen species or inhibit their reaction with biological structures. The destruction of most free radicals and activated oxygen species relies on the oxidation of endogenous antioxidant mainly scavenging and reducing molecules.

In the present study, PbAc orally administrated to mice did not affect markedly body and kidneys weights. This may probably due to the absence of effect on food intake by mice as well as tissues growth promoting. However, *B. vulgaris* co-treatment increased significantly body weight. These finding are congruent with those of Rajain et al[23], who indicates a significant positive effect of *B. vulgaris* on weight gain in chicken.

Moreover, lead exposure increased significantly blood and kidneys lead burden. This finding corroborates other studies of lead-induced toxicity in animals[24-25]. Co-administration of BV extract significantly reduced blood and renal intracellular lead levels in a dose-dependent manner. Previous studies reported that *B. vulgaris* contains polyphenolics[8,26], which are known for their metal-chelating ability[27-28]. This might be one of the mechanisms by which *B. vulgaris* decreases lead content in blood and kidney, and therefore potentiates its body clearance.

On the other hand, urea and creatinine are considered...
**TABLE 2: PLASMA LEVELS OF CREATININE, UREA AND URIC ACID OF MICE**

<table>
<thead>
<tr>
<th>Parameters</th>
<th>CT</th>
<th>PbAc</th>
<th>PbAc+25</th>
<th>PbAc+50</th>
<th>PbAc+100</th>
<th>PbAc+150</th>
</tr>
</thead>
<tbody>
<tr>
<td>Creatinine (mg/dl)</td>
<td>1.16±0.03</td>
<td>1.9±0.03*</td>
<td>1.85±0.02#</td>
<td>1.84±0.04#</td>
<td>1.66±0.03#</td>
<td>1.17±0.02*</td>
</tr>
<tr>
<td>Urea (mg/dl)</td>
<td>43.26±1.98</td>
<td>75.9±1.54#</td>
<td>73.46±1.08#</td>
<td>71.5±2.53#</td>
<td>54.14±0.72*#</td>
<td>45.96±1.34*#</td>
</tr>
<tr>
<td>Uric acid (mg/dl)</td>
<td>2.34±0.11</td>
<td>1.26±0.03#</td>
<td>1.27±0.05*</td>
<td>1.84±0.04*#</td>
<td>1.93±0.05*#</td>
<td>2.21±0.05*</td>
</tr>
</tbody>
</table>

Plasma levels of creatinine, urea and uric acid of mice control group (CT), lead acetate-treated mice (PbAc), and *Berberis vulgaris*-treated mice (0, 25, 50, 100, 150 mg/kg) in combination with PbAc (PbAc+BV). Values are mean±SEM, for five animals in each group. *Values differ significantly (P<0.05) from normal control group (CT). #Values differs significantly (P<0.05) from toxic control group (PbAc).

**TABLE 3: GSH, GPX, CAT, SOD, MDA AND PCO CONTENTS IN KIDNEYS OF MICE**

<table>
<thead>
<tr>
<th>Parameters</th>
<th>CT</th>
<th>PbAc</th>
<th>PbAc+25</th>
<th>PbAc+50</th>
<th>PbAc+100</th>
<th>PbAc+150</th>
</tr>
</thead>
<tbody>
<tr>
<td>GSH (µmol/g wet tissue)</td>
<td>4.31±0.23</td>
<td>2.6±0.11*</td>
<td>2.68±0.33*</td>
<td>2.9±0.24*</td>
<td>3.54±0.2*</td>
<td>4.04±0.34*</td>
</tr>
<tr>
<td>GPx (nmol GSH/min/mg protein)</td>
<td>63.9±0.28</td>
<td>40.12±1.89#</td>
<td>41.69±2.36*</td>
<td>39.24±1.66*</td>
<td>55.13±3.87*</td>
<td>61.14±5.19*</td>
</tr>
<tr>
<td>CAT (µmol H₂O₂/min/mg protein)</td>
<td>37.45±0.61</td>
<td>23.43±1.6#</td>
<td>23.46±1.48#</td>
<td>26.45±0.82#</td>
<td>33.31±0.81*</td>
<td>34.79±1.92*</td>
</tr>
<tr>
<td>SOD (U/mg protein)</td>
<td>15.37±0.4</td>
<td>10.37±1.04#</td>
<td>10.19±0.35#</td>
<td>11.88±0.53#</td>
<td>12.73±0.56*</td>
<td>14.48±1.04*</td>
</tr>
<tr>
<td>MDA (nanomoles of MDA/g of tissue)</td>
<td>36.42±1.64</td>
<td>59.83±3.36*</td>
<td>62.37±2.94*</td>
<td>60.67±2.27*</td>
<td>41.63±4.00*</td>
<td>36.33±2.97*</td>
</tr>
<tr>
<td>PCO (nmol carbonyl/mg protein)</td>
<td>0.69±0.06</td>
<td>1.43±0.16*</td>
<td>1.375±0.09*</td>
<td>1.33±0.09*</td>
<td>1.17±0.10*</td>
<td>0.98±0.06*</td>
</tr>
</tbody>
</table>

GSH, GPx, CAT, SOD, MDA and PCO contents in Kidneys of mice control group (CT), lead acetate-treated mice (PbAc), and *Berberis vulgaris*-treated mice (0, 25, 50, 100, 150 mg/kg) in combination with PbAc (PbAc+BV). Values are expressed as mean±SEM for five animals in each group. *Values differs significantly (P<0.05) from normal control group (CT). #Values differ significantly (P<0.05) from toxic control group (PbAc).

Fig. 1: Histograms of kidney sections.

Histograms (∼20) of kidney sections of normal control mice (A), Pb-acetate-treated mice (B), PbAc+25 group (C), PbAc+50 group (D), PbAc+100 group (E), PbAc+150 group (F); The black arrows indicated normal glomerular structure. The red arrows showed degenerated glomerulus.
as biomarkers of the first and advancing stages of kidney damage, respectively[29]. In this study, PbAc-treated mice showed increase in plasma creatinine and urea levels, which may reflect the renal failure onset[30]. Kidney plays a major role in the clearance and biotransformation of metals[31], and therefore is considered as a primary target organ for metals induced toxicity[32]. PbAC could increase blood urea by more than one mechanism, including enhancement of proteins catabolism, conversion of ammonia to urea by induction of arginase-enzyme synthesis[33], and inhibition of amino acids incorporation in proteins[34]. The hyperuricemia induced by PbAc treatment might result from over-production and/or reduced renal excretion of uric acid[35], and elevation of endogenous oxygen species levels[36]. Co-treatment with BV extract inhibited PbAc-induced nephrotoxicity, as indicated by significant restoration of serum creatinine, urea and uric acid.

Administration of PbAc decreased significantly reduced glutathione (GSH). This is consonant with some reports asserting that lead induces GSH level depletion in Pb-acetate treated rats[37]. Oxidative stress occurs when generation of free radicals exceed the capacity of antioxidant defense mechanisms. Lead toxicity provokes free radical-induced cellular damage through two main mechanisms: the production of ROS such as O$_2^-$, H$_2$O$_2$, and peroxy radical, and the direct decline of antioxidant reserves[37]. GSH acts in conjunction with antioxidant enzymes in the decomposition of H$_2$O$_2$ and other organic hydroperoxides. In this context, depletion of intracellular GSH is a prognostic indicator of increased cytotoxicity of H$_2$O$_2$ in kidney cells[38-39]. Therefore, BV aqueous extract could mitigate oxidative stress by increasing GSH level through scavenging H$_2$O$_2$, as it was demonstrated by Gurer and Ercal[4]. BV polyphenols might be the potential agents in the chelating processes[40-41].

This study also elucidated that PbAc-treated mice exhibited decreased kidneys antioxidant enzymes activities such as Glutathione peroxidase (GPx), catalase (CAT), and superoxide dismutase (SOD). These enzymes are metalloproteins that detoxify peroxides (–OOH), H$_2$O$_2$, and O$_2^-$, respectively[42]. These antioxidant enzymes depend on essential trace elements and prosthetic groups for their structure and activity, and therefore are the potential targets of lead toxicity[43].

Therefore, the decreased activities of these antioxidant enzymes observed in lead-treated mice indicated a failure of antioxidant defense system to overcome the influx of ROS induced by PbAc exposure. Furthermore, many reports depicted the inhibition of the antioxidant enzymes activities as the main mechanism of lead-induced cytotoxicity[44-45]. B. vulgaris administration increased antioxidant enzyme activities in PbAc-treated mice, which could be induced by lowering free radicals generation. In this concern, B. vulgaris extract could react with free radicals or with lipid peroxidation metabolites, and may also enhance tissue thiol contents, which lead finally to the reduction of oxidative modification of enzymes, and enhancement of antioxidants and glutathione-related enzymes activities. In this context, the beneficial antioxidative effect of B. vulgaris has been previously reported in some animal models[46-47].

In the current study, PbAc induced the increase of kidneys malondialdehyde (MDA) and protein carbonyls (PCO) levels in the toxic control group. Lead is known to have toxic effects on membrane structure and functions[48]. Hence, altered lipid and proteins composition of membranes due to lead exposure is associated with an increase in the concentration of MDA, and PCO that may result in altered membrane integrity, permeability, and function. These would increase the susceptibility to lipid peroxidation and generation of free radicals[48-49]. The increased lipid peroxidation and oxidative modification of enzymes contents, which are assessed by increase in PCO levels after Pb-acetate exposure, indicate the implication of oxidative stress in lead-induced nephrotoxicity. Co-administration of B. vulgaris restored MDA and PCO levels respectively to normal values. Therefore, it seems likely that the B. vulgaris-extract effect is mediated by scavenging free radicals and decreasing hydroxyl radical generation. These findings are congruent with previous reports that concern the antioxidant effects of the whole plant or its main component, berberine[41-41].

Mice chronic exposure to lead was associated with noticeable modifications of the biochemical parameters and significant alteration of the kidney architecture, which was confirmed by histopathological assessment. In fact, the renal histoarchitecture of the lead-treated mice presented signs of nephrotoxicity, such as glomerulus cells degeneration characterized by glomerular hypercellularity. However, the higher dose (150 mg/kg) of the B. vulgaris-extract improved significantly the histopathological alterations and restored the glomerular structure near to normal status. Similar histopathological changes were found by Dewanjie et al.[24], who reported that the histological
evaluation of the kidneys of lead-intoxicated rats revealed severe histopathological changes, and treatment with *Corchorus olitorius* restored the tissue architecture almost similar to their normal ultrastructure.

In conclusion, the current work demonstrates that *B. vulgaris* has a nephroprotective effect against lead-induced toxicity in the kidney of mice. The mechanisms of effectiveness involve the quenching of free radicals, antioxidant and metal chelating ability of the plant extract. Therefore, *B. vulgaris* extract appears to be a promising agent for protection against lead-induced nephrotoxicity. Dietary supplementation with *B. vulgaris* could be a useful method to protect populations at high risk of environmental and/or occupational lead chronic intoxication, as this medicinal plant is widely used as food in many countries.

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Nil.

**Conflicts of interest:**

There are no conflicts of interest.

**REFERENCES**


