
Antibacterial Screening of N-arylamino-3-chloro-4-(4'-Dimethyl Amino) Phenyl-Azetidin-2-Ones

PRATIBHA SHARMA*, PRITI INDAPURKAR AND ANUPAM MANDLOI
Institute of Chemical Sciences
Devi Ahilya University, Indore - 452001.

A number of new N-arylamino-3-chloro-4-(4'-dimethylamino) phenyl-azetidin-2-ones have been synthesised and their structures have been established on the basis of consistent elemental, IR, spectral data. Antibacterial activity has been performed using agar diffusion technique involving paper disc method against *E. coli* (445), *Pseudomonas diminuta* (MTCC 1609) and *Bacillus subtilis* (MTCC 441). All the synthesised azetidines-2-one have shown significant antibacterial activity. It has been observed that N-(4'-nitro) phenylamino-3-chloro-4-(4'-dimethylamino) phenyl azetidines-2-one is found to be more potent against *E. coli*. Molecular refractive index (M_R) correlates linearly to the drug activity with correlation coefficient of 0.99.

AZETIDINES and their derivative have been extensively explored for their applications in the field of medicine^{1,2}. Likewise, azetidines-2-ones are of great importance because of the use of β -lactam derivative as antibacterial agents³. Recent years have witnessed a great upsurge in the treatment of tuberculosis⁴. Keeping this generalization in view, we report herein a new method of synthesis and antimicrobial activity of N-arylamino-3-chloro-4-(4'-dimethylamino) phenyl-azetidines-2-ones.

EXPERIMENTAL

All melting points are uncorrected. IR spectra were scanned in KBr on Shimadzu 460 IR spectrophotometer. All the chemicals used were of AR grade. N-arylamino-3-chloro-4-(4'-dimethylamino) phenyl azetidines-2-ones were synthesised and characterized in the following manner.

1-Arylhydrazono-1-(4"-dimethylamino) phenylmethanes [B] (0.02 M) obtained by condensation of corresponding hydrazines [A] with p-dimethyl-amino benzaldehyde was added to a constant stirred solution of 1,4-dioxan (40 ml), triethylamine (0.02 mol) and

chloroacetyl chloride (0.02 mol). The reaction mixture was mechanically stirred at 50°. The reaction vessel was kept at room temperature for 30 min and then refluxed for 8 h. On cooling the precipitate so obtained was filtered off, thoroughly washed with water and dried. The product was recrystallised from dimethylformamide (DMF) as white crystals. Yield 65%.

Anal of Ia: IR bands (in cm^{-1}) at 3400 (-NH), 3070 (=C-H), 2860 (-CH) 1660 (C=C), 1740 (C=O, characteristic of 4 membered carbonyl), 1600, 1500, 860, 740 (aryl).

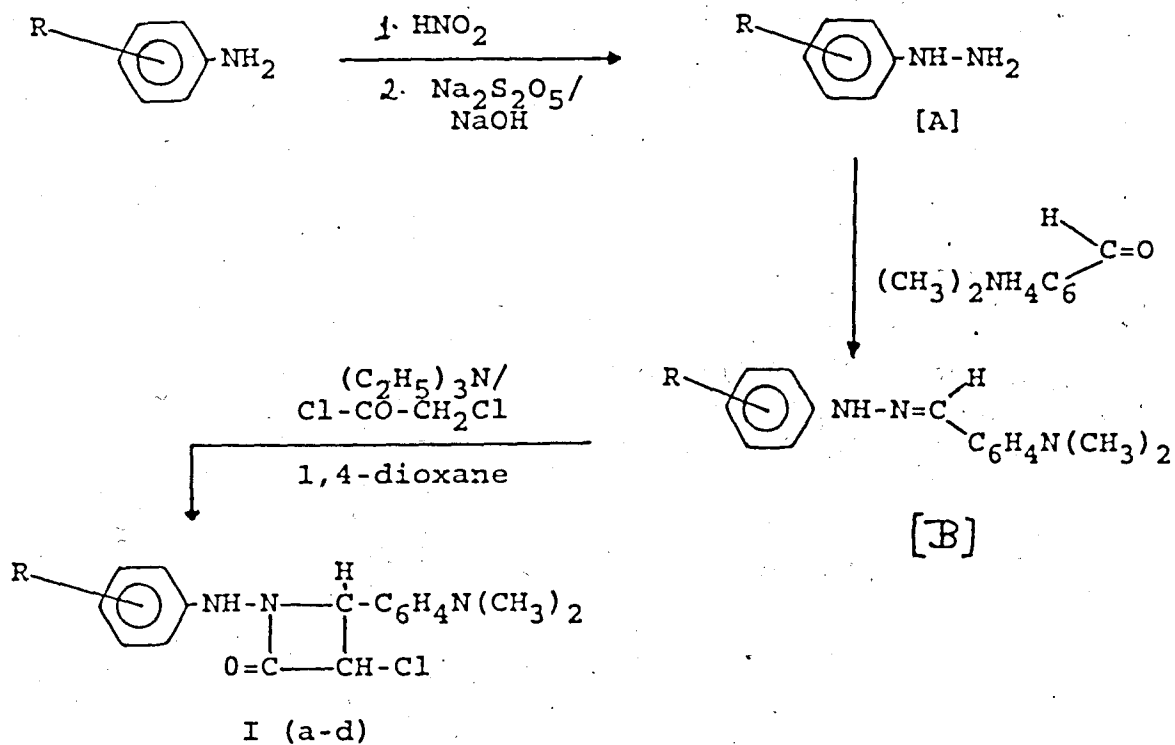
¹NMR (CDCl₃ + DMSO-d₆) ppm: 1.4 (s, 3H, CH₃), 2.4 (H, CHC₆H₅), 3.2-3.4 (d, 1H, CHC₆H₅), 7.5-7.65 (m, Ar-H), 10.2 (H,NH, C₆H₅).

METHODS

General Method for Antibacterial Screening: The method employed for antibacterial activity was the agar diffusion technique⁵ involving paper disc method. Bacterial cultures used for the study were *E. coli* (445), *Pseudomonas diminuta* (MTCC 1609) and *Bacillus subtilis* (MTCC 441).

*For correspondence

An overview of their synthetic pathway is depicted in Scheme-1



R =

-H, -NO₂, -Cl, -CH₃,

SCHEME 1

Antibacterial Screening

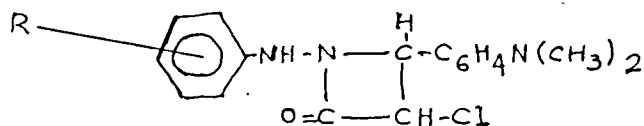
All the bacterial strains were maintained on nutrient agar slants⁶. Culture was inoculated in peptone water and incubated overnight. All the synthesised compounds were screened for antibacterial activity at 0.25 - 1.00 m/M100 ml. all the screening results were evaluated in terms of the zone of inhibition in cm. The percent inhibition has been calculated using the formula; % inhibition = $(\alpha - \beta) / \alpha \times 100$, where α and β stands for inhibition zone of control drug (Sulphamethoxole has been used as the control drug) and inhibition zone of azetidino-phenylamine derivatives, respectively.

RESULTS AND DISCUSSION

A systematic perusal of data presented in Table-1 reveals that maximum inhibition is found against *E. coli* and in the series N-(4'-nitro) phenylamino-3-chloro-4-(4'-dimethylamino) phenyl azetidino-2-one is found to be a more potent antibacterial agent.

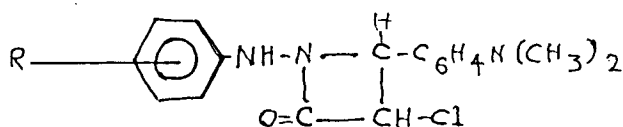
To study quantitative structure activity relationships (QSAR) molecular refractive index (M_R) has been calculated by the method of Dreishach⁷ and reported in Table 2. A perusal of data reveals that M_R is correlated linearly to the drug activity [A]⁸ for activity against *E. coli*, the equation is $[A] = 0.45 + 0.92 \times M_R$ with correlation of coefficient of 0.992.

Table 1 : Antibacterial activity of N-arylamino-3-chloro-4-(4' dimethylamine) phenyl azetidion-2-ones



Compound	LD ₅₀ mg/Kg	Organism	control drug αcm	Azetidin -2-one cm ^β	% inhibition α-β/α x 100
Ia	18.10	<i>P. dimunita</i>	2.2	2.3	-04.54
		<i>B. subtilis</i>	2.8	2.5	+10.71
		<i>E.coli</i>	2.4	2.8	-16.66
Ib	20.14	<i>P. dimunita</i>	2.2	2.9	-31.81
		<i>B. subtilis</i>	2.8	2.9	-03.57
		<i>E.coli</i>	2.4	3.3	-37.50
Ic	14.14	<i>P. dimunita</i>	2.2	2.7	-22.72
		<i>B.subtilis</i>	2.8	2.2	-21.42
		<i>E.coli</i>	2.4	3.5	-45.83
Id	25.18	<i>P. dimunita</i>	2.2	3.3	-50.00
		<i>B. substilis</i>	2.8	2.9	-03.57
		<i>E.coli</i>	2.4	2.5	-04.16

Table 2 : Characteristic data of N-arylamino-3-chloro-4-(4' dimethylamine) phenyl azetidion-2-ones



Compd	R	M _r	M.P °C	Yield %	C Calc % (found %)	H	N
Ia	H	39.100	244	62	64.86 (64.32)	5.40 (4.94)	13.35 (13.00)
Ib	-NO ₂	45.309	249	67	56.58 (55.64)	4.71 (4.70)	15.53 (15.43)
Ic	-Cl	43.976	248	60	56.04 (55.67)	4.67 (4.70)	15.38 (15.38)
Id	-CH ₃	43.727	246	56	65.5 (65.45)	6.06 (6.00)	12.74 (12.72)

Procedure For The Determination of LD₅₀

Before screening for above activities, substituted azetidin-2-one compounds were subjected to acute toxicity studies to find out LD₅₀. During which, 10 rats were considered in each group and were observed for 72 h. According to the toxicity studies, the LD₅₀ value and the magnitude of toxicity are inversely related to each other i.e. the smaller the LD₅₀ value, the more toxic the chemical. All the synthesised compound did not show any toxicity upto a dose of 8.1 mg / kg in rats. LD₅₀ for each synthesised compound is depicted in the table 1.

ACKNOWLEDGEMENTS

We thank the Director, Institute of Chemical Sciences, DAVV, Indore for providing all the facilities to carry out this

work, and to University Grants Commission, New Delhi for supporting this research project.

REFERENCES

1. Werber, G, Buccheri, F, Nato, R, and Salozzo. P, *J. Heterocyclic Chem*, 1978, 15, 1537.
2. Stephenton, J. S, Vorella, E, Micromastarras, E. D, and Alexander, N. J. *Heterocyclic chem*, 1978, 16, 1373.
3. Koppel, G. A., In; "Small Ring Heterocycles", Wiley Intersciences, New York, 1983.
4. Vashi, B. S, Mehta, D. S, and Shah, V. H, *Ind J Chem*, 1995, 34B, 802.
5. Paper, R. P, "The actinomycetes", 1901, 88, 12.
6. Cavanagh F, In; "Analytical microbiology", Academic Press, New York, 1963, 726.
7. Dreishach RR, In; "Physical properties of chemical compounds", Am. Chem Soc, Washington D. C. 1961.
8. Barlow RB, In; "Quantitative Aspects of Chemical Pharmacology". Croom Helm, London, 1980.