New Spectrophotometric Methods for the Determination of Racecadotril in Bulk Drug and Capsules

T. VETRICHELVAN* AND S. PRABAKARAN

Department of Pharmaceutical Analysis, Adhiparasakthi College of Pharmacy, Melmaruvathur - 603 319, India.

Two simple and sensitive spectrophotometric methods (A and B) for the determination of racecadotril in bulk drugs and pharmaceutical formulations are described. In method A, methanol was used as solvent and shows absorption maximum at 231 nm. In method B, the solvent used was acetonitrile:water in the ratio of 1:3 and shows absorption maximum at 232 nm. The Beer's law range for method A is 25-100 μ g/ml and 20-80 μ g/ml for method B. When capsules dosage forms were analyzed, the results obtained by the proposed methods are in good agreement with the labeled amounts and the results were validated statistically.

Racecadotril is an effective and safe drug for acute diarrhea in adults and children. Chemically racecadotril is N-[(R,S)-3-acetylmercapto-2-benzylpropanoyl]-glycine benzyl ester¹⁻³ (fig. 1). The Drug Controller General of India approved it as an antidiarrheal in October 2001⁴. It is not yet official in any Pharmacopoeia. A survey of literature revealed that a few HPLC⁵ methods were reported for the estimation of racecadotril in biological fluids. In the present report, the paper describes two simple, economical and sensitive spectrophotometric methods for the determination of racecadotril in bulk samples and solid dosage forms. In method A, methanol was used as solvent and in method B, acetonitrile:water (1:3) was employed.

Absorbance measurements were made on a Shimadzu-1700 double beam UV/Vis spectrophotometer. All the solvents used for analytical studies of racecadotril were of analytical reagent grade. Pharmaceutical grade of racecadotril was kindly gifted by M/s. Micro Labs,

*For correspondence E-mail: vetrisel@yahoo.com Pondicherry, India. The capsules of racecadotril used for the studies were procured from a local Pharmacy. The solubility of racecadotril was determined in a variety of solvents using essentially a method of Schefter and Higuchi⁶. From the solubility studies, methanol and acetonitrile:water (1:3) were selected as solvents for UV spectroscopical studies of racecadotril in bulk drug and capsules dosage form. The λ max was determined in methanol and acetonitrile:water (1:3).

Herital of and the total volume was brought to 100 mil with

Fig. 1: Structure of the racecadotril

In method A, racecadotril (25 mg) was dissolved in methanol and the total volume was brought to 100 ml with

methanol. It was further diluted to obtain 25-100 μ g/ml with methanol. The absorbance was measured at 231 nm against methanol as blank. The calibration curve was plotted in the concentration range of 0.025 to 0.1 mg/ml of racecadotril in methanol. In method B, racecadotril (50 mg) was dissolved in acetonitrile:water (1:3) and the total volume was brought to 100 ml. It was further diluted to obtain 20-80 μ g/ml with acetonitrile:water (1:3). The absorbance was measured at 232 nm against reagent as blank. The calibration curve was plotted in the concentration range of 0.02-0.08 mg/ml of racecadotril.

In method A, twenty capsules of each formulation S_1 and S₂ containing racecadotril were powdered, weighed equivalent to 25 mg of racecadotril and dissolved with 20 ml methanol, vigorously shaken for 20 minutes and filtered through Whatmann filter paper No. 41. Repeated the extraction three times, filtered and made up to 100 ml with methanol. The dilutions were made in the same manner as described under bulk drug. The absorbance measurements were made six times for each formulation. The amount of racecadotril was calculated from the respective calibration curve (method A). In method B, Capsules powder equivalent to 50 mg of racecadotril was weighed and extracted with successive quantities of acetonitrile:water (1:3), filtered through Whatman filter paper No. 41 and made up to 100 ml with the same solvent. Subsequent dilutions and absorbance were

TABLE 1: OPTICAL (CHARACTERISTICS	OF PRC	POSED
METHODS	0.1	YA	

Parameters	Method A	Method B
λmax (nm)	231	232
Beers law limit (µg/ml)	25-100	20-80
Sandell's sensitivity ^a (µg/cm ² /0.001 A.U)	0.087412	0.077519
Molar extinction coefficient (1 mol ⁻¹ cm ⁻¹)4.411×10 ³	5.282×103
Correlation coefficient (r ²)	0.999329	0.999623
Regression equation (y=mx+c)	0.566	0.514
Slope (m)	0.011148	0.01299
Intercept [©]	0.014737	0.007121
Confidence limit with 0.05 level (95%)	± 0.5800	± 0.3965
Confidence limit with 0.01 level (90%)	± 0.9096	± 0.6218

^aSandell sensitivity (S) = $10^{-3}/a$; S = Number of micrograms of the determined per ml of a solution having a cross section of 1 cm² and absorbance of 0.001 and a = absorbance of 1 µg/ml solution determined in a cuvette with an optical path length of 1 cm. measured as described under procedure for calibration curve. The amount of racecadotril was calculated from the respective calibration curve (Method B). Recovery experiments were performed by adding six different quantity of drug in previously analyzed sample, but with in the limit of Beer's Law amount. The percentage of drug recovered was calculated by a mathematical relation followed by Sane *et al*⁷.

The range of linearity for racecadotril was determined in methanol and acetonitrile:water and found to be 25-100 μ g/ml and 20-80 μ g/ml respectively. Beer's Law limits⁸, molar absorptivity, Sandell's sensitivity9, slope and intercept of regression analysis using least square method, precision and accuracy¹⁰ of the analysis are summarized in Table 1. The results of pharmaceutical preparations (capsules) containing racecadotril are shown in Table 2. The outcome of recovery studies revealed the method B is more sensitive and accurate than the method A. As an additional check on the accuracy of the methods, recovery experiments were performed by adding known amount of pure drug to pre- analyzed dosage forms. The percentage of drug recovered (99-102%) was good agreement with the added amount and labeled claim. Recovery experiments indicated the absence of interference from commonly encountered pharmaceutical additives and excipients. The proposed methods were validated statistically¹¹ and found reproducible results. All statistical data proves validity of the proposed methods, which can be applied in industries for routine analysis of this method to analyze racecadotril in bulk drug and Pharmaceutical preparation. These results indicate that the proposed methods are sensitive, accurate, precise and reproducible.

ACKNOWLEDGEMENTS

The authors are thankful to Arulthiru Bangaru Adigalar, President, Thirumathi Lakshmi Bangaru Adigalar, Vice-President and Dr.T.Ramesh, Managing Director, -ACMEC- Trust, Melmaruvathur for providing necessary facilities to carry out this work.

TABLE 2: ASSAY AND RECOVERY STUDIES OF RACECADOTRIL IN CAPSULE DOSAGE FORM

Pharmaceutical formulation [#]	Labeled amount (mg/ capsule)	····· · · · · · · · · · · · · · · · ·		Percent recovery	
	Method A (mg) (mean±SD)	Method B (mg) (mean±SD)	Method A	Method B	
Capsule - S ₁	100	100.3±0.1469	99.9±0.077	99.11	100.45
Capsule - S ₂	100	100.1±0.2104	100.19±0.0194	99.08	101.65

*Mean of six determinations. "The commercial preparations used were, Capsule - S, is Redotil, and Capsule- S, is Cadotril.

www.ijpsonline.com

REFERENCES

- Budavari, S., Eds., In; The Merck Index, 13th Edn., Merck Research Laboratories, Division of Merck and Co., Inc., Whitehouse Station, NJ., 2001, 1450.
- Anonymous, CIMS, BIO-GARD (P) Ltd., Bangalore, January 2005, 88, 172.
- Reynolds, J.E.F., Eds., In; Martindale, The Extra Pharmacopoeia, 33rd Edn., The Pharmaceutical Press, London, 2002, 1337.
- 4. Nagpal, J. and Gogia, S., Indian Pediatrics, 2004, 41, 218.
- 5. Zhao, Z. and Liu, SC., Chin. Pharm. J., 2001, 22, 267.
- 6. Schefter, E. and Higuchi, T., J. Pharm. Sci., 1963, 52, 781.
- 7. Sane, R.J., Smita G. J., Mary, F., Aamer R.K. and Premangsus, H., Indian Drugs, 1999, 36, 317.

- Beckett. H.A. and Stenlake B.J., Practical Pharmaceutical Chemistry, 4th Edn., CBS Publishers, New Delhi, 2001, 275.
- 9. Sandell E.B., Colorimetric determination of traces of metals, Inter Science, NewYork, 1950, 29.
- Kamboj, P.C., Pharmaceutical Analysis, 1st Edn., Vallabh Publications, New Delhi, 2003, 1, 155.
- Gupta, S.C., Fundamentals of Statistics, 4th Edn., Himalaya Publishing House, New Delhi, 1999.

Accepted 15 April 2007 Revised 23 August 2006 Received 10 June 2005

Indian J. Pharm. Sci., 2007, 69 (2): 307-309