-Research Paper -

Phenolic Constituents with Antioxidant and Antiviral Activities from *Phyllanthus urinaria* Linnea

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Liang et al.: Phenolic Constituents with Antioxidant and AntiHBV Activities from Phyllanthus urinaria

Phyllanthus urinaria Linnea is commonly used in folk medicine with considerable therapeutic utility. Sixteen phenols of *Phyllanthus urinaria* were isolated from ethanol extract and their structures were identified by spectroscopic data. Oxygen radical absorbance capacity, radical scavenging assay and HBsAg and HBeAg ELISA kits were used to evaluate antioxidant and antiviral activities. All the isolates showed significant antioxidant activity. Especially, isolates 4 and 5 had the oxygen radical absorbance capacity of 5.12 and 8.13 U/mol and strong radical scavenging activity with IC_{50} values of 5.50 and 5.55 μ M, respectively. Additionally, these studies also demonstrated that the compounds 1, 2, 11, 12, and 16 inhibited secretion of HBsAg and HBeAg in HepG2.2.15 cells. This study suggested that the phenolic constituents of *Phyllanthus urinaria* possessed antioxidant activities and these might play an important role in antihepatitic effects.

Key words: *Phyllanthus urinaria* Linnea, isolation and identification, phenolic constituents, antioxidant activity, antihepatitis B activity

Phyllanthus urinaria Linnea (Euphorbiaceae) is an important traditional herb, which is widely used in subtropical area of Asia for liver protection, hepatitis, nephrolithiasis, dropsy, jaundice, and diabetes^[1]. It has attracted more and more attention due to its ability to treat HBsAg carriers reported by Thyagarajan *et al.*^[2].

Contemporary pharmacological research reported that P. urinaria exhibited antioxidant activity and significantly reduced nitric oxide radicals associated with cell-mediated immune diseases^[3]. Geraniin, a tannin from P. urinaria, was reported to exhibit semicarbazide-sensitive amine oxidase inhibitory, antihypertensive activities^[4]. antioxidant. and Additionally, ethanol, methanol and acetone extracts of P. urinaria inhibited HSV-2 infection in vitro^[5]. Ellagic acid, a phenolic molecule identified in P. urinaria, was reported to exhibit antihepatitis B virus activity through down-regulating hepatitis B virus (HBV) gene expression and decreasing the secretion of HBsAg and HBeAg in HepG 2 2.2.15 cell line^[6]. Phyllanthin, phyltetralin, trimethyl-3,4dehydrochebulate. methylgallate, rhamnocitrin. methyl brevifolincarboxylate, β-sitosterol-3-O-β-D-

glucopyranoside, quercitrin, and rutin were isolated from the ethanol extract of this plant for searching potent natural antiinflammatory agents, and their antioxidant, and antiinflammatory activities *in vitro* were assayed^[7]. Recently, (4R,6S)-2-dihydromenisdaurilide (DHMD) as an efficient inhibitor of hepatitis C virus (HCV) entry as well as (–)-menisdaurilide was isolated from the plant, and its mechanism on antiHCV infection was investigated *in vitro*^[7].

Earlier phytochemical studies on *P. urinaria* indicated that its major constituents were tannins, flavonoids, lignans, coumarins, and polyphenols^[8-12]. Among these compounds, a few compounds such as gallic acid, rutin, ellagic acid, and quercetin^[13-16] have sufficiently been studied. Some of small molecule polyphenols have also been confirmed possessing significant bioactivities, and their pharmacological activities may mainly stem

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from their free radical scavenging and their actions on cell signaling pathways^[17,18].

Although a series of chemical constituents and their activities from *P. urinaria* had been revealed earlier, further systematic phytochemical investigations and evaluations of biological activities are necessary for uncovering the fundamental constituents responsible for these therapeutic effects. The present study was aimed to investigate the phenol compounds from the ethanol extract of *P. urinaria* for potential antioxidant and antiviral activities. Isolation of 16 phenol compounds and the antioxidant and antiHBV activities exhibited by these are presented herein.

MATERIALS AND METHODS

High performance liquid chromatography (HPLC) was performed on an Agilent 1260 HPLC system equipped with an diode array detector, using column A (Ultimate XB-C18, 5 µm, 4.6×250 mm, Welch, Potamac, MA, USA) for analysis and column B (Ultimate XB-C18, 5 µm, 10×250 mm, Welch, Potamac, MA, USA) for semi-preparative HPLC purification. Open column chromatography (CC) was conducted on silica gel (200-300 mesh, Haiyang Chemical Group Corporation, Qingdao, China) and Sephadex LH-20 (25-100 mm, Pharmacia, Uppsala, Sweden). HSGF254 silica gel thin-layer chromatographic (TLC) plates (Qingdao Marine Chemical, China) were used to chemical analysis of fractions. The spraying reagent used for TLC detection was 10 % H₂SO₄ in ethanol. All nuclear magnetic resonance (NMR) spectra were carried out on Bruker AV-300 spectrometer (Bruker Instrument, Inc, Zurich, Switzerland). High-resolution electrospray ionization mass spectrometry (HR-ESI-MS) was achieved on an Agilent 6210 LC/MS TOF mass spectrometer (Agilent Technologies, Santa Clara, CA, USA). UV spectra were recorded on a Jasco V-550 UV/Vis spectrometer (Jasco Corporation, Tokyo, Japan). The optical rotations were recorded on a Jasco digital polarimeter (Jasco Corporation, Tokyo, Japan) using a thermostable optical glass cell.

Fluorescein sodium salt, 2,2'-azobis (2-methylpropionamidine) dihydrochloride (AAPH), 6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid (Trolox) were purchased from Macklin (Shanghai, China). Dulbecco's modified Eagle's medium (DMEM), heat-inactivated fetal bovine serum (FBS), penicillin A and streptomycin were purchased from Gibco (USA). Thermo Scientific Multiskan GO Microplate Spectrophotometer (Thermo Scientific, USA) and GENios Microplate System and Magellan work station (Tecan, Switzerland) were used for oxygen radical absorbance capacity (ORAC) and 2,2-diphenyl-1-picrylhydrazyl (DPPH assays). All the other reagents and solvents were provided by Tianjin Damao Chemical Company (Tianjin, China).

Plant material:

The plant material was provided by Shenzhen Hospital Affiliated to Guangzhou University of Chinese Medicine, Shenzhen, China, and authenticated in the Department of Pharmacognosy, College of Pharmacy, Jinan University. A voucher specimen (no. 20110901) was also deposited in the department.

Extraction and isolation:

Ten kilograms of the dried and ground material of *P. urinaria* was soaked in 95 % ethanol and extracted by percolation method. The extract (900 g) was obtained from the soaked solution after evaporation of the solvent. The extract was suspended in distilled water and partitioned successively with petroleum ether, chloroform, and ethyl acetate (EtOAc). The EtOAc extract (130 g) was subjected to silica gel CC eluting with CHCl₃–CH₃OH (100:0 to 0:100) in gradient to yield 8 fractions after combining based on TLC patterns. Various chromatographic (TLC, silica gel, Sephadex LH-20, ODS and HPLC) and spectroscopic methods (NMR, MS, UV) were used for the isolation and structural identification of compounds.

Cell lines and cell culture:

HepG2.2.15 cells (a HBV-transfected human HepG2 cell line) and HepG2 cells were cultured in DMEM medium with 10 % FBS and 1 % antibiotic mixture containing 100 units/ml penicillin A and 100 units/ml streptomycin in a humidified incubator with 5 % CO_2 at 37°.

Antihepatitis B virus effect in vitro:

HepG2.2.15 cells $(2 \times 10^4 \text{ cells/well})$ were seeded in 24-well plates. After 48 h incubation, cells were treated with the absence or presence of various concentrations of samples for 6 d, and the culture medium was refreshed every 3 d by fresh drug containing medium. Meanwhile, the supernatants on day 3 and day 6 were harvested to determine the concentration of HBsAg and HBeAg using an ELISA kit. Lamivudine (3TC) was used as positive control and the absence of sample was served as blank control. Each test was carried in

triplicate, and the HepG2 cells were used for testing cell viability using MTT assay.

Antioxidant activity test:

ORAC assay was applied for evaluating the antioxidant activity of each compound^[19,20]. Briefly, 20 µl of 10 µM samples to be tested or 20 µM Trolox, dissolved in phosphate-buffered saline (PBS), were transferred to each well. One hundred and forty microliters of final concentration of AAPH and 20 µl of final concentration of 63 nM fluorescein sodium salt were rapidly added to the reaction system in the dark. Fluorescence was immediately read at 538 nm (λ_{ex} : 485 nm). The decrease of fluorescence values were recorded every 2 min at 37° with 5 s stirring before measurement until no fluorescence was detected in sample. The total volume of the reaction system was supplied to 200 µl by PBS and the Trolox was served as internal control. The area under the curve (AUC) was calculated using the Eqn. below, AUC = $2 \times (f_0 + f_1 + \ldots + f_{n-1} + f_n) - f_0 - f_n$, where f_0 is the initial fluorescence at 0 min, and f_n is the fluorescence read at corresponding time. The net AUC is equal to the difference value between the AUC of the blank and each sample. ORAC values were calculated as, ORAC value = (net AUC_{sample} /net AUC_{Trolox})×(molarity of Trolox/molarity of sample).

DPPH assay:

The radical scavenging activity was determined using the DPPH method in a microtiter plate^[20,21]. Twenty microlitres of various concentrations of test samples diluted in methanol (0-30 µM final concentration) and 180 µl of 150 µM DPPH solution in methanol were added to 96-well microtiter plate. The mixture was incubated for 30 min in the dark with shaking before readings were taken. Quercetin served as a positive control, and sample was substituted by methanol in the blank. For each test, the loss on the absorbance was measured, and the DPPH radical scavenging rate (S %) was calculated in the following Eqn., S % = (Abs $_{blank}$ -Abs $_{sample}$)/Abs $_{blank}$, where Abs $_{sample}$ and Abs $_{blank}$ are the absorbance of the DPPH radical solution incubated with or without the tested sample. For each compound, four concentrations were evaluated.

All the biological assays on antihepatitis B virus effect *in vitro*, antioxidant activity test with ORAC and DPPH methods were done from 2016 to 2017.

Statistical analysis:

All values were presented as means±SD. One-May-June 2019 Indian Journal of P way ANOVA was used to evaluate the data statistic differences using Prism 6 software. At least three independent tests were performed for all assay condition.

RESULTS AND DISCUSSION

Sixteen phenolic compounds were obtained from P. urinaria extract, their structures were identified by spectroscopic methods including ¹H, ¹³C NMR, and ESI-MS, and comparison with the data reported in previous studies (fig. 1). These were identified to be emodin-8-O- β -D-glucopyranoside (1)^[22], catechin (2), epicatechin (3), chebulanin (4)^[23], albibrissinoside В $(5)^{[24]}$ $(-)-(2R)-1-O-\beta$ -D-glucopyranosyl-2- $\{2$ methoxy-4-[(E)-formylvinyl]phenoxyl}propane-3-ol $(6)^{[25]}$, (+)-pinoresinol $(7)^{[26]}$, 3,4,8,9,10pentahydroxydibenzo[b,d]pyran-6-one $(8)^{[27]}$ brevifolin (9)^[28], gallic acid (10)^[29], 3-O-methylgallic acid (11)^[30], ethyl gallate (12)^[29], ferulic acid (13)^[31], caffeic acid (14)^[32], *p*-coumaric acid (15)^[33], protocatechuic acid (16)^[32]. All of these compounds except isolate 10 were isolated from this plant for the first time. These compounds are all phenols due to the presence of phenol group (OH) in their structures, and most of them are polyphenols with two or more phenolic OHs.

The antiviral activity against hepatitis B virus of compounds 1-2, 4-7, 10-12, and 16 were investigated *in vitro* using ELISA assay, and the results were summarized in Table 1. The phenols tested in experiment showed potential antiHBV activity with stronger capacity against HBsAg secretion than HBeAg secretion at the third day or sixth day. Among them, compounds 1, 2, 11, 12 and 16 were the most active ones. They significantly inhibited HBsAg and HBeAg secretion at the maximum noncytotoxic concentration and showed stronger inhibition than the positive control lamivudine (3TC).

Antioxidant activities of isolates 1-12 and 15-16 were tested using the ORAC assay. According to the result of ORAC assay (Table 2), the isolated compounds 2, 4, 5, 11 and 15 exhibited obvious antioxidant activities with the ORAC values of 4.54, 5.12, 8.13, 4.85 and 4.88 U/mol. The curves of their relative fluorescence intensity were shown in fig. 2, and they were even stronger than the positive control quercetin. Compound 5 was the strongest antioxidant among them.

The main mechanism of phenolic antioxidants is free radical scavenging. These isolates were evaluated using



Fig. 1: Structures of the compounds 1-16

	TABLE 1: A	ANTIHEPATITIS B	VIRUS	ACTIVITIES	OF	COMPOUND
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Compound	CC_0 in HepG 2	Concentration (µM)	Inhibition of	f HBsAg (%) ^a	Inhibition of HBeAg (%) ^a		
Compound			Day 3	Day 6	Day 3	Day 6	
1	>100	100	28.41±3.30	41.20±0.82	11.87±1.00	22.19±6.06	
2	>100	100	14.82±4.33	30.80±2.34	4.17±0.10	45.82±3.84	
4	66.61±2.35	50	-	8.09±1.00	-	18.60±1.09	
5	>100	100	-	8.90±3.58	-	26.05±5.43	
6	>100	100	-	5.61±1.81	-	17.22±1.88	
7	>100	100	-	11.56±3.62	-	17.71±1.23	
10	106.93±5.29	100	1.22±0.21	12.63±0.19	10.83±1.51	15.48±1.57	
11	109.69±1.39	100	20.45±2.87	26.13±0.61	10.44±1.89	24.88±0.36	
12	26.75±2.86	25	21.69±2.32	30.95±3.82	3.93±0.21	26.48±5.75	
16	>100	100	20.68±8.26	21.73±4.16	4.88±0.44	32.17±1.75	
3TC ^b	-	500	17.26±1.97	22.14±2.17	6.73±0.33	12.31±2.11	

'-'Means no test because of the limit of amounts; the compounds 3, 8, 9, 13, 14 and 15 are no tested as their limit of amounts; 'a'mean percent inhibition values±SD (n = 9); 'b'used as positive control

the DPPH radical scavenging assay with quercetin as the positive control. Results of the scavenging activity were demonstrated by the IC_{50} values, which were presented in Table 2. Additionally, the fig. 3 implied that compounds 4, 5, 9, 10, 11 and 12 indicated strong DPPH radical scavenging activity.

As a traditional antihepatitis herb in Eastern Asia, *P. urinaria* is rich in polyphenol compounds as reported in various phytochemical investigations earlier^[4+11]. These investigations included the quantitative and

qualitative analysis of phytochemical ingredients such as the location and confirmation of active extracts^[34-36], and isolation of some compounds, even new natural products^[4,5,9-11] from this plant. It is worthy to report that research involving bioactivity-guided purification of *P. urinaria* led to the discovery of a HCV entry inhibitors loliolide^[37] and DHMD^[7]. Furthermore, ellagic acid from *P. urinaria*, which effectively suppressed the HBeAg secretion in HepG2 2.2.15 cells as a novel antiHBV compound^[7]. These studies showed

TABLE	2:	ANTIOXIDAN	T	ACTI\	/ITIES	OF
COMPOU	NDS	MEASURED	BY	ORAC	AND	DPPH
ASSAYS						

Compound	ORAC (U/mol) ^a	IC ₅₀ (μΜ) ⁶
1	2.30±0.02	>50
2	4.54±0.03	>50
3	1.22±0.04	-
4	5.12±0.29	5.50±0.09
5	8.13±0.23	5.55±0.08
6	0.82±0.02	>50
7	2.50±0.15	>50
8	1.90±0.09	-
9	2.06±0.10	10.16±0.16
10	2.73±0.02	6.25±0.16
11	4.85±0.11	13.26±0.17
12	2.91±0.12	6.14±0.10
15	4.88±0.12	>50
16	2.63±0.18	32.28±0.10
Quercetin ^c	2.59±0.14	8.64±0.10

""Mean ORAC values±SD (n=9); ""the concentration of tested compounds required to scavenge 50 % of DPPH radicals; 'c'used as positive control. Note: U was equivalent Trolox of 1 mol; '-' means no test because of the limit of amounts; the compounds 13 and 14 were not tested due to limited availability



Fig. 2: Fluorescence curves in the presence of compounds 2, 4, 5, 11 and 15

(**→→**) AAPH-; (**—■→**) AAPH+; (**—▲→**) Trolox; (**—*****—**) compound 2; (----) compound 4; (-----) compound 5; (---+---) compound 11; (-I-) compound 15

that P. urinaria contained many complex and diverse constituents with potent biological activity, especially the phenolic compounds, which exhibited various biological activities such as antioxidant^[18], antiviral^[38], antitumor^[39].

Our previous investigations on P. urinaria reported isolation of a series of flavonoid glycosides with antioxidant activities^[12]. In this study, 16 phenolic compounds were isolated from P. urinaria, most of which contained two or more phenol groups. In In ORAC and DPPH assay, all compounds nearly exhibited different levels of ORAC and radical scavenge capacity, in which most of the tested compounds exhibited strong ORAC and the DPPH scavenging IC_{50} values were determined using dose-response curves. ORAC, DPPH assay demonstrated that these phenols exhibited good antioxidant activities. Among these polyphenols, the phenolic acids 3-O-methylgallic acid (11) showed significant activity both in the antioxidant and antivirus assays compared to other phenolic acids such as gallic acid (10), ethyl gallate (12) and protocatechuic acid (16). As to the antiviral activity, previous studies showed the gallate structure played a critical role in the inhibition of HBV activity^[40]. In the antiHBV assay, the 3-O-methylgallic acid (11) and protocatechuic acid (16), especially the ethyl gallate (12) exhibited remarkable antiviral capacity.

Oxidative stress can result from increased production of reactive oxygen species (ROS). To maintain a proper redox balance, excrescent ROS may be scavenged, and the activities of oxidant enzyme may be diminished^[41]. The antioxidant activities of polyphenols mainly stem from their ROS scavenging on the oxidative stress injury. Previous studies revealed that the effect of HBV on the host antioxidant system and Fas-signalling is a crucial factor in the development of hepatocellular carcinoma, because the change of antioxidant enzyme



Fig. 3: Dose-response curves in the DPPH radical scavenging assay

-) Quercetin; (—=—) compound 4; (—•—) compound 5; -) compound 9; (-----) compound 10; (--+---) compound 11; (—**•**—) compound 12; (—*—) compound 16

system sensitized the HBV transgenic mice to the Fas signaling pathway^[42]. Furthermore, the ROS levels mediated by HBV X proteins were down-regulated by the treatment with antiviral drugs^[43].

Therefore, the antioxidant activities of phenolic compounds from *P. urinaria* may relate to the antiHBV activities, such as the compounds 2, 4, 5 and 11 not only shown strong antioxidant activity but also significant antiviral activity. Besides, among these polyphenols, 3-O-methylgallic acid (11) showed prominent biological actions both in antioxidant and antiviral activities comparing with the gallic acid (10), ethyl gallate (12) and protocatechuic acid (16). These finding suggested that the phenols from *P. urinaria* may play an important role in antihepatitis activities and the antiHBV effect of *P. urinaria* may relate to its antioxidative activities.

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