

# Vitamin K<sub>2</sub>-4 and K<sub>2</sub>-7 Estimation in Nutraceutical Solid Dosage Forms by Post Column Derivatization with Fluorescence Detection

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**Karuppiyah and Anver Basha: Analytical Estimation of Vitamin K<sub>2</sub>-4 and Vitamin K<sub>2</sub>-7**

A post column derivatization method for the estimation of fat soluble vitamins K<sub>2</sub>-4 and K<sub>2</sub>-7 was developed by reverse phase HPLC and validated as per ICH guidelines. The compounds were extracted by solvent extraction with acetone followed by evaporation in a rotary evaporator. The residue was dissolved in ethanol and injected in to a chromatograph consisting of C18 column (Waters symmetry, 150×4.6, 3.5 μ) and fluorescence detector. The derivatization reagent was prepared by dissolving 136 mg of zinc chloride, 40 mg of sodium acetate and 0.1 ml of glacial acetic acid in methanol. The mobile phase comprises of methanol:isopropyl alcohol:acetonitrile:zinc chloride buffer solution 850:90:50:10 with a flow rate of 1 ml/min. The overall percentage recoveries of five different levels were found to be 99.85 and 100.5%, respectively. The linearity of the analytical method was determined from 10% to 120% level and the linear regression coefficients were 0.9991 and 0.9995 which is well within the acceptance criteria of 0.999. The limits of detection and limits of quantification were determined based on signal to noise ratio. The established values were 0.050, 0.50 μg/ml and 0.005, 0.047 μg/ml which is much lesser than available literature limits. The developed method can be more suitable for the estimation of K<sub>2</sub>-4 and K<sub>2</sub>-7 present in drug substances as well as drug product formulations.

**Key words:** Post column derivatization, nutraceutical solid dosage forms, estimation of K<sub>2</sub>-4 and K<sub>2</sub>-7 by HPLC, zinc chloride buffer solution, analytical estimation

Vitamin K<sub>2</sub> includes several subtypes; the two subtypes most studied are menaquinone-4 (menatetrenone, MK-4) and menaquinone-7 (MK-7). Vitamin K<sub>2</sub>-4 (fig. 1a) is yellow coloured powder, molecular formula C<sub>31</sub>H<sub>40</sub>O<sub>2</sub>, molecular weight 444.65 g/mol, soluble in acetone and ethyl alcohol<sup>[1]</sup>. Vitamin K<sub>2</sub>-7 (fig. 1b) is light yellow microcrystalline powder, molecular formula C<sub>46</sub>H<sub>64</sub>O<sub>2</sub>, molecular weight 648.99 g/mol, soluble in petroleum ether, acetone, ethyl alcohol and insoluble in water<sup>[1]</sup>. Vitamin K is used for the treatment of anticoagulant-induced prothrombin deficiency caused by warfarin<sup>[2-4]</sup>. The American Academy of Paediatrics recommends that vitamin K should be given to all new born babies as a single intramuscular dose to prevent vitamin K deficiency bleeding. In Japan, vitamin K<sub>2</sub> is used for the management of osteoporosis. The fermented soya product *nattō* is rich in menaquinone-7<sup>[2-4]</sup>.

Derivatization reaction with the reduction of zinc metal ions to isolate compounds of closely eluting peaks by chromatography is fast emerging tool for analytical

separation. The dosage form formulated with two active ingredients such as K<sub>2</sub>-4 and K<sub>2</sub>-7 along with excipients is very challenging effort to separate from its moiety. In order to separate the K series vitamins with almost same structural properties, the post column derivatization technique with fluorescence detection was used. The reduction mechanism of components with zinc chloride derivatization reagent present in mobile phase was interacted with column stationary phase to achieve separation<sup>[5-13]</sup>.

The literature survey for K<sub>2</sub>-4 and K<sub>2</sub>-7 reveals that no independent method available for simultaneous and efficient separation of these two vitamins. The present study reveals that nutraceutical formulation with

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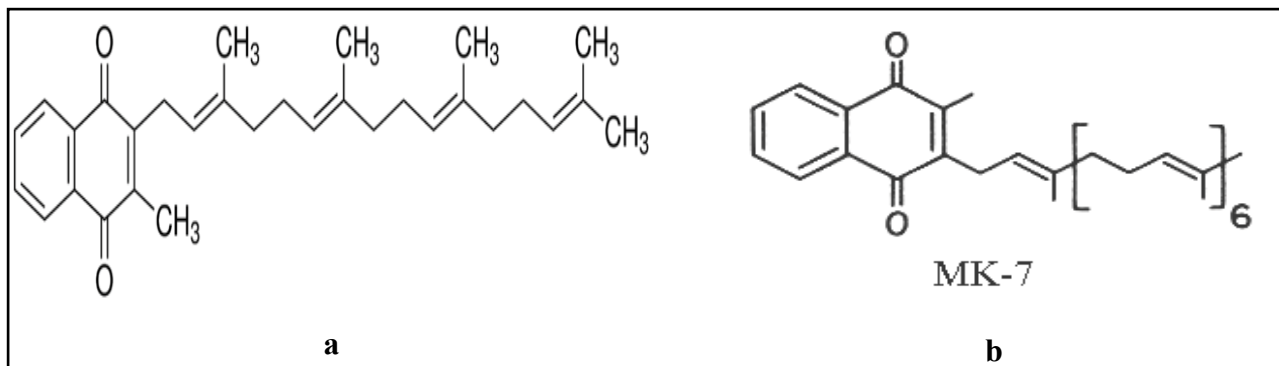


Fig. 1: Structures of analytes.

a. menaquinone K<sub>4</sub> or vitamin K<sub>2</sub>-4; b. menaquinone K<sub>7</sub> or vitamin K<sub>2</sub>-7.

vitamins was chosen to develop analytical separation method using Soxhlet's extraction with acetone solvent followed by evaporation with a rotary evaporator and fluorescence detection<sup>[14-17]</sup>. The developed method is precise, linear, accurate, robust and stability indicating compared to other UV detection regular HPLC methods. The method for estimation of K<sub>2</sub>-4 and K<sub>2</sub>-7 from its tablets formulation is more reliable for routine analysis. The developed method was validated as per International Conference on Harmonisation (ICH) guidelines<sup>[18-20]</sup>.

## MATERIALS AND METHODS

K<sub>2</sub>-4 and K<sub>2</sub>-7 standards were received from M/s Anthem Cellutions Pvt. Ltd., Bangalore, India were certified the purity 2100 µg/g and 2500 µg/g, respectively. Acetone, methanol and isopropyl alcohol were purchased from Merck Chemicals, Mumbai, India.

Zinc chloride, sodium acetate and glacial acetic acid were acquired from Rankem, Avantor Performance India Limited, Thane, India. The analytical grade ethanol (99.9% purity) was procured from England. The ultra-pure water generated from MilliQ water purification system was used for analysis. The nutraceutical formulation tablets containing K<sub>2</sub>-4 and K<sub>2</sub>-7 (Maxical Plus, Lifestar Pharma Pvt. Ltd, Mankind Group Company, New Delhi) was purchased from local market and used for this study as marketed formulation. Shimadzu LC-2010 CHT model liquid chromatographic system consists of a quaternary pump, auto-sampler with cooler, column oven, variable wave length detector (UV/Vis) and fluorescence detector (RF-20Axs, Shimadzu Corporation, Japan). The analysis was controlled by LC solution/lab solution software 1.25 version. The analytical balance (Mettler Toledo JB1603-C/FACT), ultrasonic bath (Citizen HIC-CP-4820), Soxhlet's extraction flask (Merck),

Rotary evaporator (Buchi) and pH meter (Eutech pH tutor) were used for this study.

### Chromatographic conditions and test solution preparation:

Transfer an accurately weighed quantity of 100 mg each of K<sub>2</sub>-4 (purity 2100 µg/g) and K<sub>2</sub>-7 (purity 2500 µg/g) in to cotton thimble loosely plugged with cotton and placed it in a Soxhlet's extractor. The round bottom flask fitted with extractor and water condenser was placed on heating mantle with a temperature controller. The extractor was filled with acetone solvent and carried out the extraction procedure for 30 min at about 70°. This operation was repeated for 2 more times. The extract was collected in a glass beaker and transferred in to a rotary evaporator collection flask for evaporation at 80° under vacuum till to get the residue. The obtained residue was dissolved in ethyl alcohol and transferred quantitatively in to a 50 ml volumetric flask and make up the volume with ethyl alcohol. The prepared solution was filtered through 0.45 µm nylon membrane filter.

The marketed formulation tablets were triturated in to fine powder using mortar and pestle. Transfer an accurately weighed quantity of powdered tablets equivalent to 100 mg each of K<sub>2</sub>-4 and K<sub>2</sub>-7 in to cotton thimble loosely plugged with cotton and placed it in a Soxhlet's extractor. The round bottom flask fitted with extractor and water condenser was placed on heating mantle with a temperature controller. The extractor was filled with acetone solvent and the extraction procedure was carried out for 30 min at about 70°. This operation was repeated for 2 more times. The extract was collected in a glass beaker and transferred in to a rotary evaporator collection flask for evaporation at 80° under vacuum till to get the residue. The obtained residue was dissolved in ethyl alcohol and transferred

quantitatively in to a 50 ml volumetric flask and make up the volume with ethyl alcohol. The prepared solution was filtered through 0.45  $\mu\text{m}$  nylon membrane filter.

The quantification method of two pharmaceutical ingredients present in tablet formulation was conducted by HPLC analysis with fluorescence detection. The mobile phase comprises of methanol, isopropyl alcohol, acetonitrile and zinc chloride solution (0.1 M) in the ratio of 850:90:50:10. The pump flow rate of 1 ml/min. Injection volume of 50  $\mu\text{l}$ , auto sample cooler temperature 10°, the excitation wavelength of 248 nm and emission wavelength 430 nm were set as chromatographic conditions. The peaks of interest were separated in an isocratic elution by using Waters symmetry C18 (150 $\times$ 4.6 mm, 3.5  $\mu\text{m}$ ) column. K<sub>2</sub>-4 and K<sub>2</sub>-7 were reduced by fluorescent derivative present in mobile phase and then detected by fluorescence at an excitation wavelength of 248 nm and an emission wavelength of 430 nm.

#### Method validation:

The analytical method was validated as per ICH guidelines. The parameters selected for validation are precision, linearity, accuracy, limits of detection (LOD), limits of quantification (LOQ), robustness and specificity to check for placebo interference.

The precision was performed by determining six times assay preparations of K<sub>2</sub>-4 and K<sub>2</sub>-7 at working concentration level. The method precision percentage relative standard deviation (%RSD) of six times assay determinations should be less than 2%. The ruggedness or intermediate precision was performed by repeating method precision with different chemist, different system, different day and different analytical column of same manufacturer.

The linearity of an analytical method was established by plotting calibration curves against concentration versus area response. The calibration curve was generated by prepared stock solution of each pharmaceutical active ingredient in to seven different concentrations (10-120%) of 0.5, 1.0, 1.5, 2.5, 3.5, 5.0 and 6.0  $\mu\text{g/ml}$ . The linear regression coefficient of each component should be greater than 0.999.

The components recovery of the method was carried out by spiking 5 different concentrations (25%, 50%, 100%, 150% and 200%) with placebo. The tablet placebo was spiked with K<sub>2</sub>-4 and K<sub>2</sub>-7 at five different levels of concentration. For K<sub>2</sub>-4, the spiked concentration levels were 1.2, 2.3, 4.7, 7.0 and 9.3  $\mu\text{g/ml}$ . For K<sub>2</sub>-7,

the spiked concentration levels were 1.3, 2.5, 5.0, 7.5 and 10.1  $\mu\text{g/ml}$ . The recovered concentrations of each active component should be between 98 to 102% and %RSD should be not more than 2%.

LOD is a qualitative limit which has been established by serial dilution of samples prepared in extraction method till to get the minimum concentration of signal to noise ratio greater than 3.0. The stock solution concentration of 5  $\mu\text{g/ml}$  was prepared by spiking the components with placebo. The determined LOD value for K<sub>2</sub>-4 and K<sub>2</sub>-7 were 0.005 and 0.05  $\mu\text{g/ml}$ , respectively.

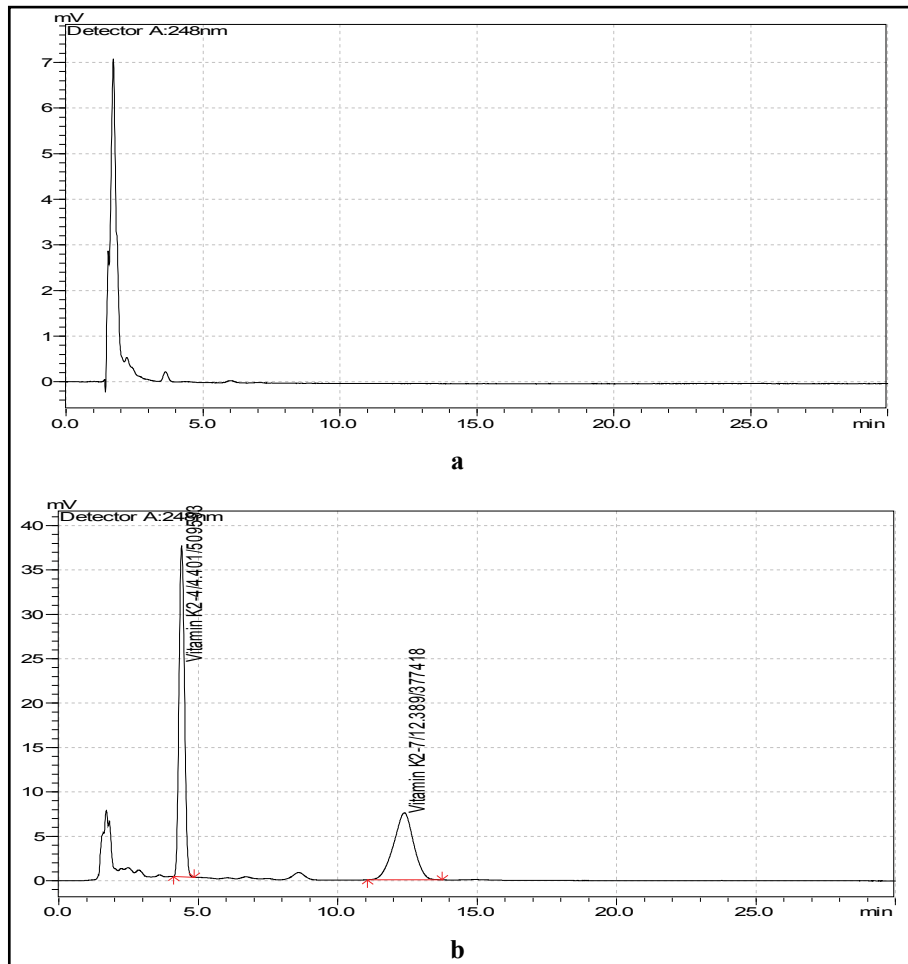
LOQ is the quantitative limit which has been determined by serial dilution of samples prepared in extraction method till the minimum concentration of signal to noise ratio is greater than 10. The stock solution concentration of 5  $\mu\text{g/ml}$  was prepared by spiking K<sub>2</sub>-4 and K<sub>2</sub>-7 with placebo. The determined LOQ values for K<sub>2</sub>-4 and K<sub>2</sub>-7 were 0.047 and 0.50  $\mu\text{g/ml}$ , respectively.

The robustness of the method was performed by deliberately modified variations of chromatographic conditions such as flow rate, column temperature, mobile phase ratio and pH of buffer concentrations. The results obtained were compared with method precision results and %RSD is not more than 2%.

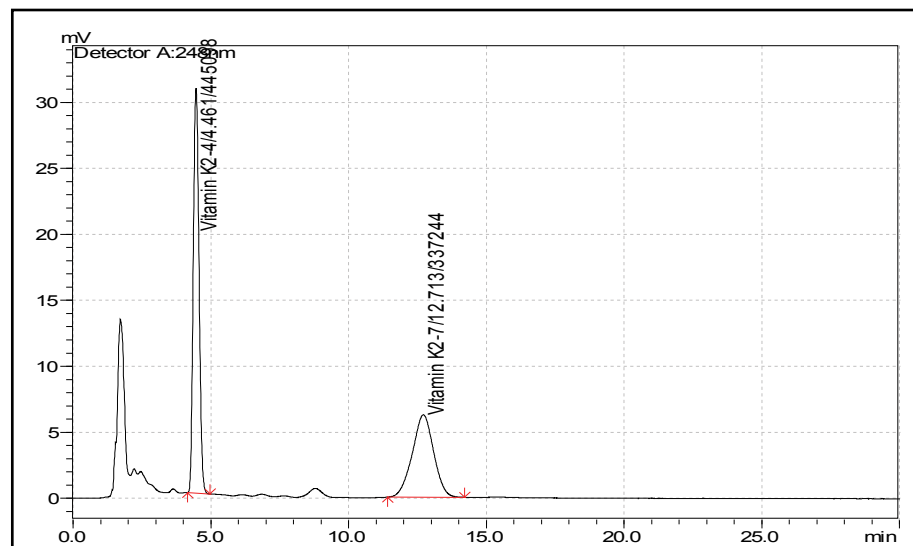
The tablet formulation placebo (all the excipients except active components) was prepared and chromatographed with the similar condition as applied for assay determination. The active components (K<sub>2</sub>-4 and K<sub>2</sub>-7), blank solution, placebo solution (same concentration as standard solution) were chromatographed and examined for peak interference. The major components peak purity was verified by using diode array detector (DAD).

## RESULTS AND DISCUSSION

The post column derivatization method for simultaneous estimation of two vitamins present in tablet formulation by solvent extraction followed by fluorescence detection was developed and validated. The mobile phase comprising of methanol:isopropyl alcohol:acetonitrile:zinc chloride solution (850:90:50:10), at 1 ml flow rate with column temperature 40° was optimised to give better resolved peaks with system suitability parameters for K<sub>2</sub>-4 and K<sub>2</sub>-7 (fig. 2). The retention times (RT) for K<sub>2</sub>-4 and K<sub>2</sub>-7 were 4.4 and 12.7 min, respectively (fig. 3). In direct sample preparation method the components were



**Fig. 2: Chromatograms of blank and analytes.**  
Chromatogram for blank solution preparation (a) and standard solution preparation of K<sub>2</sub>-4 and K<sub>2</sub>-7 (b).



**Fig. 3: Chromatogram for nutraceutical tablets preparation.**  
Chromatogram for nutraceutical tablets preparation containing K<sub>2</sub>-4 and K<sub>2</sub>-7.

not extracted completely from matrix components and hence the recovery for estimation of drug from drug product formulation is not satisfied. But by present

study the solvent extraction method was used to recover components completely with suitable precision and accuracy. In order to increase the recovery of components

the sample preparation techniques were optimized by running different amount of sample, increasing number of extraction time, temperature and different kinds of solvents. The system suitability conditions such as standard solution RSD, peak asymmetry, resolution between two peaks and theoretical plates were well within the limits of acceptance criterion. Based on the above study extraction methodology was optimised to get more recovery of analytes from matrix components present in tablet formulation.

The developed method was validated as per ICH guidelines for analytical method validation. The repeatability of method was established by six times assay determination of sample preparation with working concentration. The average percentage of assay values were 100.4% of K<sub>2</sub>-4 (99.7, 101.4, 100.9, 99.5, 100.3 and 100.6%) and 100.7% of K<sub>2</sub>-7 (101.6, 100.8, 101.3, 100.5, 99.9 and 100.1%), respectively. The %RSD of six determinations was found to be 0.718 and 0.662%, respectively. The linearity of the analytical method was determined from 10% level to 120% level. The linear regression coefficient for K<sub>2</sub>-4 and K<sub>2</sub>-7 were 0.9991 and 0.9995, respectively which is well within the acceptance criteria of 0.999. The regression analysis data of the calibration curves is described in Table 1.

The analyte recovery was performed for 5 different levels from 25 to 200% of working concentration level. The mean recovery obtained for each level is described in Table 2. The overall percentage recovery of all levels were 99.85 and 100.5%. The %RSD were 1.15% and 1.09%, respectively. The extended lower and higher level of recovery concentrations can be justified since it was an extraction procedure. LOD

and LOQ was determined based on signal to noise ratio. The determined LOD and LOQ values were 0.005, 0.047 µg/ml and 0.05, 0.500 µg/ml respectively, which is much lesser than available literature limits. The robustness of method was performed by slightly altering the deliberately modified variations like flow rate, column temperature, mobile phase ratio and buffer pH. The results obtained was compared with method precision results and the RSD between two results is described in Table 3. The tablet placebo was prepared by mixing sodium starch glycolate, talc, polyvinyl pyrrolidone (PVPK30), magnesium stearate, aerosil, cross carmellose sodium, microcrystalline cellulose powder (MCCP102), hydroxypropyl methylcellulose (HPMC E15), titanium dioxide, polyethylene glycol (PEG 6000), isopropyl alcohol and methylene chloride in proper proportions. The peak purity of two analyte peaks and placebo interference were verified and found satisfactory results.

The system suitability parameters were described in Table 4. The developed method was found to be sensitive and selective for the determination of K<sub>2</sub>-4 and K<sub>2</sub>-7 present in combination without any placebo interference. The developed method was applied to the determination of K<sub>2</sub>-4 and K<sub>2</sub>-7 in their combined nutraceutical tablet formulation.

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**TABLE 1: LINEAR REGRESSION ANALYSIS DATA FOR CALIBRATION CURVES**

Parameters	K <sub>2</sub> -4	K <sub>2</sub> -7
Linearity range (µg/ml)	0.46-5.47	0.50-6.0
r <sup>2</sup>	0.9991	0.9995
Slope	0.10871	0.74677
Intercept	0.21808	0.16804
Average RSD of estimation	0.97473	1.1245

K<sub>2</sub>-4, K<sub>2</sub>-7, SD is standard deviation (n=3), RSD is relative standard deviation

**TABLE 2: SUMMARY OF RECOVERY STUDIES AT DIFFERENT CONCENTRATION**

Parameters	K <sub>2</sub> -4	K <sub>2</sub> -7
Spiked concentration(µg/ml)	1.17,2.33,4.66,6.99 and 9.32	1.26,2.51,5.03,7.54 and 10.06
Recovered concentration(µg/ml)	1.16,2.36,4.64,7.10 and 9.42	1.28,2.49,5.00,7.44 and 10.10
Average recovery (%)	99.1,101.3,99.6,101.6 and 101.1	101.6,99.2,99.4,98.7 and 100.4
Standard deviation	0.305,0.462,0.487,0.158 and 0.589	0.644,0.311,0.354,0.517 and 0.460
Relative standard deviation (%)	0.308,0.45,0.489,0.156 and 0.583	0.640,0.314,0.356,0.524 and 0.459

K<sub>2</sub>-4, K<sub>2</sub>-7, Five different recovery concentrations (n=3)

**TABLE 3: SUMMARY OF ROBUSTNESS STUDIES FOR K<sub>2</sub>-4 AND K<sub>2</sub>-7**

Parameters	Average assay (%)		Relative standard deviation (%)	
	K <sub>2</sub> -4	K <sub>2</sub> -7	K <sub>2</sub> -4	K <sub>2</sub> -7
Flow rate (Lower 0.8 ml)	103.1	101.5	1.88	0.56
Flow rate (Higher 1.2 ml)	102.5	100.9	1.46	0.14
Column temperature (Lower 38°)	101.9	100.4	1.05	0.21
Column temperature (Higher 42°)	102.1	100.2	1.20	0.35
Mobile phase ratio (Lower)	101.9	99.9	1.06	0.57
Mobile phase ratio (Higher)	102.4	101.1	1.41	0.28
Buffer pH (Lower)	102.0	101.3	1.13	0.42
Buffer pH (Higher)	102.8	102.1	1.68	0.99

K<sub>2</sub>-4, K<sub>2</sub>-7, Four different variables were studied at lower and higher level (flow rate, column temperature, mobile phase ratio and buffer pH)

**TABLE 4: SUMMARY OF METHOD VALIDATION PARAMETERS**

Parameters	K <sub>2</sub> -4	K <sub>2</sub> -7
Linearity range (µg/ml)	0.46-5.47	0.50-6.0
Correlation coefficient	0.9991	0.9995
Precision (%RSD)		
Method precision (n=3)	0.718	0.662
Ruggedness (n=3)	0.977	0.868
Recovery (%)	99.85	100.5
LOD (µg/ml)	0.005	0.050
LOQ (µg/ml)	0.047	0.500
Robustness	Complies	Complies
Retention time	4.4±0.2	12.7±0.2
Resolution	----	9.03
Theoretical plates	3259	4248
Tailing factor	1.386	1.264

Each validation parameter and system suitability was described

out the work.

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### Conflicts of interest:

There are no conflicts of interest.

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